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(Z)-3-Chloro-N-[(Z)-3-(3-chloro-2-methylphenylimino)butan-2-ylidene]-2-methylaniline

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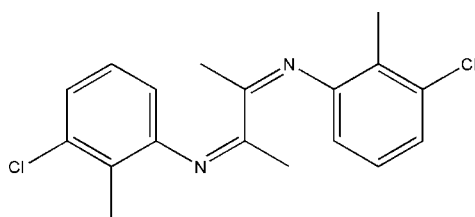
Received 14 November 2011; accepted 2 December 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.039; wR factor = 0.115; data-to-parameter ratio = 15.6.

In the title compound, $\text{C}_{18}\text{H}_{18}\text{Cl}_2\text{N}_2$, the complete molecule is generated by the application of C_2 symmetry. The $\text{C}=\text{N}$ bond has an E configuration. The dihedral angle between the benzene ring and the 1,4-diazabutadiene plane is 66.81 (9°).

Related literature

For background to the applications of the olefin polymerization Ni(II)- α -diimine catalysts, see: Johnson *et al.* (1995); Killian *et al.* (1996). For the effect of the ligand structure on the activity of the catalyst and properties of the products, see: Popeney & Guan (2010); Popeney *et al.* (2011); Yuan *et al.* (2005). For related structures, see: Kose & McKee (2011); Wei *et al.* (2011).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{18}\text{Cl}_2\text{N}_2$

$M_r = 333.24$

Monoclinic, $P2_1/n$
 $a = 8.032$ (6) Å
 $b = 7.372$ (5) Å
 $c = 14.475$ (10) Å
 $\beta = 93.533$ (7)°
 $V = 855.5$ (11) Å³

$Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.38$ mm⁻¹
 $T = 296$ K
 $0.23 \times 0.21 \times 0.19$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2008)
 $T_{\min} = 0.918$, $T_{\max} = 0.932$

5279 measured reflections
 1588 independent reflections
 1199 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.115$
 $S = 1.06$
 1588 reflections

102 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.22$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.28$ e Å⁻³

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BQ2319).

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supplementary materials

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(Z)-3-Chloro-N-[(Z)-3-(3-chloro-2-methylphenylimino)butan-2-ylidene]-2-methylaniline

J. Yuan, W. Xu, T. Mei, Y. Liu and X. Wang

Comment

There is a considerable interest in the development of new late transition metal catalysts for the polymerization of α -olefins since Brookhart discovered highly active α -diimine nickel catalysts (Johnson *et al.*, 1995; Killian *et al.*, 1996). It is well known that the ligand structure had significant influence on the product properties and polymerization activities (Popeney & Guan, 2010; Popeney *et al.*, 2011; Yuan *et al.*, 2005). In this study, we designed and synthesized the title compound as a bidentate ligand, and its molecular structure was characterized by X-ray diffraction. In the solid state, the ligand exhibits a C_2 symmetry. The single bond of 1,4-diazabutadiene fragment is (E)-configured. The dihedral angle between the benzene ring and 1,4-diazabutadiene plane is 66.81 (9)°. (Figure 1.) In the crystal packing, there is no hydrogen-bond between the ligand molecules.

Experimental

Formic acid (0.5 ml) was added to a stirred solution of 2,3-butanedione (0.052 g, 0.6 mmol) and 3-chloro-2-methylaniline (0.0170 g, 1.2 mmol) in methanol (20 ml). The mixture was refluxed for 24 h, then cooled and the precipitate was separated by filtration. The solid was recrystallized from dichloromethane/cyclohexane ($v/v = 8:1$), washed with cold ethanol and dried under vacuum to give the title ligand 0.18 g (90%). Anal. Calcd. for $C_{18}H_{18}Cl_2N_2$: C, 64.87; H, 5.44; N, 8.41; Cl, 21.28. Found: C, 64.97; H, 5.33; N, 8.21; Cl, 21.59. Crystals suitable for X-ray structure determination were grown from a solution of the title compound in a mixture of cyclohexane/dichloromethane (1:2, v/v).

Refinement

All hydrogen atoms were placed in calculated positions with C—H distances of 0.93 and 0.96 Å for aryl and methyl type H-atoms. They were included in the refinement in a riding model approximation, respectively. The H-atoms were assigned $U_{iso} = 1.2$ times U_{eq} of the aryl C atoms and 1.5 times U_{eq} of the methyl C atoms.

Figures

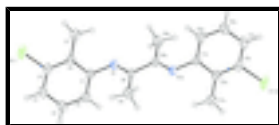


Fig. 1. Molecular structure of the title compound, using 30% probability level ellipsoids. Primed atoms are related by the symmetry code $(-x + 1, -y + 1, -z)$.

(Z)-3-Chloro-N-[(Z)-3-(3-chloro-2-methylphenylimino)butan-2-ylidene]-2-methylaniline

Crystal data

$C_{18}H_{18}Cl_2N_2$

$F(000) = 348$

supplementary materials

$$M_r = 333.24$$

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$$a = 8.032 (6) \text{ \AA}$$

$$b = 7.372 (5) \text{ \AA}$$

$$c = 14.475 (10) \text{ \AA}$$

$$\beta = 93.533 (7)^\circ$$

$$V = 855.5 (11) \text{ \AA}^3$$

$$Z = 2$$

$$D_x = 1.294 \text{ Mg m}^{-3}$$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 2237 reflections

$$\theta = 2.8\text{--}28.0^\circ$$

$$\mu = 0.38 \text{ mm}^{-1}$$

$$T = 296 \text{ K}$$

Block, colorless

$$0.23 \times 0.21 \times 0.19 \text{ mm}$$

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube
graphite

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2008)

$$T_{\min} = 0.918, T_{\max} = 0.932$$

5279 measured reflections

1588 independent reflections

1199 reflections with $I > 2\sigma(I)$

$$R_{\text{int}} = 0.034$$

$$\theta_{\max} = 25.5^\circ, \theta_{\min} = 2.8^\circ$$

$$h = -9 \rightarrow 9$$

$$k = -8 \rightarrow 8$$

$$l = -17 \rightarrow 17$$

Refinement

Refinement on F^2

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.039$$

$$wR(F^2) = 0.115$$

$$S = 1.06$$

1588 reflections

102 parameters

0 restraints

Primary atom site location: structure-invariant direct
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0428P)^2 + 0.4563P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\min} = -0.28 \text{ e \AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.4602 (2)	0.2982 (3)	0.17826 (13)	0.0373 (5)
C2	0.5314 (2)	0.3432 (3)	0.26633 (13)	0.0382 (5)
C3	0.5281 (3)	0.2084 (3)	0.33347 (14)	0.0451 (5)
C4	0.4584 (3)	0.0403 (3)	0.31793 (16)	0.0532 (6)
H4	0.4589	-0.0453	0.3651	0.064*
C5	0.3875 (3)	0.0005 (3)	0.23100 (16)	0.0532 (6)
H5	0.3391	-0.1124	0.2192	0.064*
C6	0.3887 (3)	0.1286 (3)	0.16168 (15)	0.0469 (5)
H6	0.3411	0.1013	0.1031	0.056*
C7	0.6047 (3)	0.5274 (3)	0.28511 (17)	0.0603 (7)
H7A	0.5640	0.5748	0.3412	0.090*
H7B	0.5730	0.6070	0.2346	0.090*
H7C	0.7241	0.5182	0.2916	0.090*
C8	0.5174 (2)	0.4244 (3)	0.03386 (13)	0.0383 (5)
C9	0.6332 (4)	0.2775 (3)	0.00740 (18)	0.0672 (8)
H9A	0.6605	0.2017	0.0601	0.101*
H9B	0.7334	0.3304	-0.0136	0.101*
H9C	0.5803	0.2058	-0.0414	0.101*
Cl1	0.61699 (10)	0.25335 (11)	0.44457 (4)	0.0788 (3)
N1	0.4462 (2)	0.4357 (2)	0.10980 (11)	0.0421 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0400 (11)	0.0413 (12)	0.0309 (10)	0.0048 (9)	0.0052 (8)	0.0042 (8)
C2	0.0360 (10)	0.0453 (12)	0.0334 (11)	0.0021 (9)	0.0046 (8)	0.0013 (9)
C3	0.0424 (12)	0.0634 (15)	0.0295 (11)	0.0035 (10)	0.0009 (8)	0.0075 (10)
C4	0.0594 (14)	0.0556 (15)	0.0455 (13)	-0.0024 (12)	0.0089 (11)	0.0188 (11)
C5	0.0584 (14)	0.0491 (14)	0.0527 (14)	-0.0101 (11)	0.0075 (11)	0.0077 (11)
C6	0.0527 (13)	0.0505 (14)	0.0370 (12)	-0.0034 (10)	0.0001 (9)	-0.0003 (10)
C7	0.0726 (17)	0.0586 (16)	0.0487 (14)	-0.0108 (13)	-0.0031 (12)	-0.0026 (12)
C8	0.0443 (11)	0.0404 (11)	0.0299 (10)	0.0026 (9)	-0.0003 (8)	0.0024 (9)
C9	0.0875 (19)	0.0616 (16)	0.0553 (15)	0.0312 (14)	0.0260 (14)	0.0189 (13)
Cl1	0.0917 (6)	0.1055 (6)	0.0367 (4)	-0.0069 (4)	-0.0154 (3)	0.0111 (3)
N1	0.0523 (11)	0.0433 (10)	0.0308 (9)	0.0050 (8)	0.0021 (7)	0.0058 (8)

Geometric parameters (\AA , $^\circ$)

C1—C6	1.391 (3)	C6—H6	0.9300
C1—C2	1.404 (3)	C7—H7A	0.9600
C1—N1	1.417 (3)	C7—H7B	0.9600
C2—C3	1.392 (3)	C7—H7C	0.9600
C2—C7	1.498 (3)	C8—N1	1.273 (3)
C3—C4	1.373 (3)	C8—C9	1.493 (3)

supplementary materials

C3—C11	1.751 (2)	C8—C8 ⁱ	1.500 (4)
C4—C5	1.380 (3)	C9—H9A	0.9600
C4—H4	0.9300	C9—H9B	0.9600
C5—C6	1.378 (3)	C9—H9C	0.9600
C5—H5	0.9300		
C6—C1—C2	120.64 (19)	C1—C6—H6	119.6
C6—C1—N1	120.53 (18)	C2—C7—H7A	109.5
C2—C1—N1	118.46 (19)	C2—C7—H7B	109.5
C3—C2—C1	116.2 (2)	H7A—C7—H7B	109.5
C3—C2—C7	123.0 (2)	C2—C7—H7C	109.5
C1—C2—C7	120.82 (19)	H7A—C7—H7C	109.5
C4—C3—C2	123.8 (2)	H7B—C7—H7C	109.5
C4—C3—C11	117.42 (17)	N1—C8—C9	126.12 (19)
C2—C3—C11	118.82 (18)	N1—C8—C8 ⁱ	116.1 (2)
C3—C4—C5	118.8 (2)	C9—C8—C8 ⁱ	117.8 (2)
C3—C4—H4	120.6	C8—C9—H9A	109.5
C5—C4—H4	120.6	C8—C9—H9B	109.5
C6—C5—C4	119.8 (2)	H9A—C9—H9B	109.5
C6—C5—H5	120.1	C8—C9—H9C	109.5
C4—C5—H5	120.1	H9A—C9—H9C	109.5
C5—C6—C1	120.8 (2)	H9B—C9—H9C	109.5
C5—C6—H6	119.6	C8—N1—C1	122.45 (18)
C6—C1—C2—C3	1.1 (3)	C11—C3—C4—C5	-179.88 (18)
N1—C1—C2—C3	174.16 (17)	C3—C4—C5—C6	0.4 (3)
C6—C1—C2—C7	-178.5 (2)	C4—C5—C6—C1	-0.2 (3)
N1—C1—C2—C7	-5.4 (3)	C2—C1—C6—C5	-0.6 (3)
C1—C2—C3—C4	-0.9 (3)	N1—C1—C6—C5	-173.5 (2)
C7—C2—C3—C4	178.6 (2)	C9—C8—N1—C1	-4.5 (3)
C1—C2—C3—C11	179.13 (14)	C8 ⁱ —C8—N1—C1	176.6 (2)
C7—C2—C3—C11	-1.3 (3)	C6—C1—N1—C8	-67.6 (3)
C2—C3—C4—C5	0.2 (4)	C2—C1—N1—C8	119.4 (2)

Symmetry codes: (i) $-x+1, -y+1, -z$.

Fig. 1

